

APPENDIX NO. 1

International System SI Base Units used for measuring physical magnitudes and phenomena are the following:

Table 1: SI Base Units			
Base Quantity		Base Unit	
Name	Symbol	Name	Symbol
Length	<i>l</i>	meter	M
Mass	<i>m</i>	kilogram	Kg
Time	<i>t</i>	seconds	S
Electric current	<i>I</i>	Ampere	A
Thermodynamic temperature	<i>T</i>	Kelvin	K
Amount of substance	<i>n</i>	Mole	Mol
Luminous intensity	<i>I_v</i>	Candela	Cd

Table 2: SI derived units, whose names and symbols include SI derived units with specific names and symbols

<i>Derived quantity</i>	Derived Unit SI			
	Name	Symbol	Expressed in other SI Units	Expressed by SI Base Units
Plane angle	radian ^(a)	rad	-	$\text{m} \cdot \text{m}^{-1} = 1$ ^(b)
Solid angle	steradian ^(a)	sr ^(c)	-	$\text{m}^2 \cdot \text{m}^{-2} = 1$ ^(b)
Frequency	hertz	Hz	-	s^{-1}
Force	newton	N	-	$\text{m} \cdot \text{kg} \cdot \text{s}^{-2}$
Pressure, stress	pascal	Pa	N/m^2	$\text{m}^{-1} \cdot \text{kg} \cdot \text{s}^{-2}$
Energy, work, quantity of Heat	joule	J	$\text{N} \cdot \text{m}$	$\text{m}^2 \cdot \text{kg} \cdot \text{s}^{-2}$
Power, radiant flux	watt	W	J/s	$\text{m}^2 \cdot \text{kg} \cdot \text{s}^{-3}$
Electric charge, quantity of electricity	coulomb	C	-	$\text{s} \cdot \text{A}$
Electric potential difference, electromotive force	volt	V	W/A	$\text{m}^2 \cdot \text{kg} \cdot \text{s}^{-3} \cdot \text{A}^{-1}$
Capacitance	farad	F	C/V	$\text{m}^{-2} \cdot \text{kg}^{-1} \cdot \text{s}^4 \cdot \text{A}^2$
Electric resistance	ohm	Ω	V/A	$\text{m}^2 \cdot \text{kg} \cdot \text{s}^{-3} \cdot \text{A}^{-2}$
Electric conductance	siemens	S	A/V	$\text{m}^{-2} \cdot \text{kg}^{-1} \cdot \text{s}^3 \cdot \text{A}^2$
Magnetic flux	weber	Wb	$\text{V} \cdot \text{s}$	$\text{m}^2 \cdot \text{kg} \cdot \text{s}^{-2} \cdot \text{A}^{-1}$
Magnetic flux density	tesla	T	Wb/m^2	$\text{kg} \cdot \text{s}^{-2} \cdot \text{A}^{-1}$
Inductance	henry	H	Wb/A	$\text{m}^2 \cdot \text{kg} \cdot \text{s}^{-2} \cdot \text{A}^{-2}$
Celsius temperature	degree Celsius	$^{\circ}\text{C}$	-	K
Luminous flux	lumen	lm	$\text{Cd} \cdot \text{sr}$ ^(c)	$\text{m}^2 \cdot \text{m}^{-2} \cdot \text{cd} = \text{cd}$
Luminance	lux	lx	lm/m^2	$\text{m}^2 \cdot \text{m}^{-4} \cdot \text{cd} = \text{m}^{-2} \cdot \text{cd}$
Activity (of a radionuclide)	becquerel	Bq	-	s^{-1}
Absorbed dose, specific energy (imparted), kerma	gray	Gy	J/kg	$\text{m}^2 \cdot \text{s}^{-2}$
Dose equivalent ^(d)	sievert	Sv	J/kg	$\text{m}^2 \cdot \text{s}^{-2}$
Catalytic activity	katal	kat		$\text{s}^{-1} \cdot \text{mol}$

Table 3: SI derived units, whose names and symbols include SI derived units with specific names and symbols

Derived Quantities	Derived Unit SI	
	Name	Symbol
	Dynamic viscosity	pascal second
Moment of force	newton meter	N m
Surface tension	newton per meter	N/m
Angular velocity	radian per second	rad/s
Angular acceleration	radian per second squared	rad/s ²
Heat flux density, irradiance	watt per square meter	W/m ²
Heat capacity, entropy	joule per Kelvin	J/K
Specific heat capacity, specific entropy	joule per kilogram Kelvin	J/(kg ·K)
Specific energy	joule per kilogram	J/kg
Thermal conductivity	watt per meter Kelvin	W/(m ·K)
Energy density	joule per cubic meter	J/m ³
Electric field strength	volt per meter	V/m
Electric charge density	coulomb per cubic meter	C/m ³
Electric flux density	coulomb per square meter	C/m ²
Permittivity	farad per meter	F/m
Permeability	henry per meter	H/m
Molar energy	joule per mole	J/mol
Molar entropy, molar heat capacity	joule per mole Kelvin	J/(mol ·K)
Exposure (x and γ rays)	coulomb per kilogram	C/kg
Absorbed dose rate	gray per second	Gy/s
Radiant intensity	watt per steradian	W/sr
Radiance	watt per square meter steradian	W/(m ² ·sr)
Catalytic (activity) concentration	katal per cubic meter	kat/m ³

Table 4: Units out of SI generally accepted and used				
Quantity		Unit		
Name	Symbol	Name	Symbol	Value in SI Units
<i>Time</i>	<i>t</i>	Minute hour Day	min h d	1 min = 60 s 1 h = 60 min 1 d = 24 h
<i>Plane Angle</i>	∠	Degree Minutes Seconds	° ' "	$1^\circ = (\pi/180)$ rad $1' = (1/60)^\circ$ $1'' = (1/60)'$
<i>Volume</i>	<i>V</i>	Liter	L	1 L = 1 dm ³
<i>Mass</i>	<i>m</i>	Metric tone	t	1 t = 10 ³ kg
<i>log</i>		Neper	Np	1 Np = 1
<i>ln</i>		Bel	B	1 B = (1/2)ln 10 Np

Clarification:

It is allowed the use of some units that entered deeper use such as:

- Traditional math units for measuring the angles: arch scale, arch minute and arch.
- Traditional units for measuring of civil time (minute, hour, day and year).
- Two measuring unities that are often used in the daily life: liter for volume and ton for large mass.

Two logarithmic units: neper and bel.

Table 5: Units out of SI permitted for use values of which are determined experimentally				
Quantity		Unit		
Name	Symbol	Name	Symbol	Value in SI Units
<i>Energy</i>	E	Electron volt	eV	1 eV = 1,602 18*10 ⁻¹⁹ J (approximately)
<i>Mass</i>	m	Unified atomic mass unit	u	1 u = 1,660 54*10 ⁻²⁷ kg (approximately)
<i>Length</i>	<i>l</i>	astronomic unit	ua	1 ua = 1,495 98*10 ¹¹ m (approximately)

Clarification:

It is allowed the use of some scientific units out of SI which present important constants in science such as: astronomic unit for length, atomic unit for mass or Dalton for quantity and electron and Volt for energy which are given in table 5.

Table 6: Units out of SI accepted for use in specific areas			
<i>Quantity</i>	<i>Unit</i>	<i>Symbol</i>	<i>Value in SI Units</i>
Length	Nautical mile		1 Nautical mile = 1852 m
speed	knot		1 Nautical mile per hour = (1852/3600) m/s
Linear density	texs		1 texs=10 ⁻⁶ kg/m=1mg/m
Pressure of liquids of human body	millimeter Colon with mercury	mmHg	1 mmHg=133322Pa
surface	are	a	1 a = 1 dam ² = 10 ² m ²
	hectare	ha	1 ha = 1 hm ² = 10 ⁴ m ²
Pressure	bar	bar	1 bar = 0.1 MPa = 100 kPa = 1000 hPa = 10 ⁵ Pa
Length	Ångström	Å	1 Å = 0.1 nm = 10 ⁻¹⁰ m
Section	barn	b	1 b = 100 fm ² = 10 ⁻²⁸ m ²

Clarification:

It is allowed the use of some metric and non-metric units which are in traditional use in different scientific fields. The use of these units should be understood in the way that in the case of any use the correlation of them with respective international units- SI should be given. These are:

- Nautical mile as a unit of speed and **knot** as unit of speed which is traditionally used in marine and meteorology.
- Acre and hectare units often used for surface (in Albania it s also used a lot the very old unit with Turkish origin – dulum that is 10 acres).
- Bar as a unit of pressure with all its manifolds and submultiples build according to the respective prefixes such as milibar in meteorology and kilobar in technique.
- angstrom (**ångström**) as a unit of length in physics and barn as a unit of efficient section in nuclear physics.

Table 7: Derived units of the CGS system with special names

Quantity	Name	Symbol	Value in SI Units
<i>Work energy</i>	erg	erg	1 erg = 10^{-7} J
<i>force</i>	din (dyn)	dyn	1 dyn = 10^{-5} N
<i>Viscose</i>	puaz (poise)	P	1 P = 1 dyn s/cm ² = 0.1 Pa s
	stoks (stokes)	St	1 St = 1 cm ² /s = 10^{-4} m ² /s
<i>Magnetic induction</i>	gaus	G	1 G $\approx 10^{-4}$ T
<i>The intensity of the magnetic filed</i>	oersted	Oe	1 Oe $\approx (1000/4\pi)$ A/m
<i>Magnetic flux</i>	maksuell	Mx	1 Mx $\approx 10^{-8}$ Wb
<i>Brightness</i>	stilb	sb	1 sb = 1 cd/cm ² = 10^4 cd/m ²
<i>Illumination</i>	fot (phot)	ph	1 ph = 10^4 lx
<i>acceleration</i>	gal	Gal	1 Gal = 1 cm/s ² = 10^{-2} m/s ²

Clarification:

In table 7 are given units that can be found in literature and it is preferable that readers avoid them. With the help of this table they are transformed into the SI units and after the values are allowed for use.

Table 8: Units out of SI forbidden for use

Quantity	Name	Symbol	Value in SI Units
<i>The activity of radionuclide</i>	Curie	Ci	1 Ci = $3.7 \cdot 10^{10}$ Bq
<i>Exposure of X or gama rays</i>	Roentgen	R	1 R = $2.58 \cdot 10^{-4}$ C/kg
<i>Absorption dosage of ionization</i>	Rad	rad	1 rad = 1 cGy = 10^{-2} Gy
<i>Equivalent dosage</i>	Rem	rem	1 rem = 1 cSv = 10^{-2} Sv
<i>length</i>	unit X		1 unit X = $1,002 \cdot 10^{-4}$ nm
<i>Submultiples of unit Tesla</i>	gama	γ	1 γ = 1 nT = 10^{-9} T
	jansky	Jy	1 Jy = 10^{-26} W · m ⁻² · Hz ⁻¹
<i>length</i>	fermi		1 fermi = 1 fm = 10^{-15} m
<i>mass</i>	Metric carat		1 karat metrik = 200 mg = $2 \cdot 10^{-4}$ kg
<i>pressure</i>	torr	Torr	1 Torr = (101 325/760) Pa
	Normal atmosphere	atm	1 atm = 101 325 Pa
<i>heat</i>	calorie	cal	1 cal = 4,1868 J
<i>Length</i>	mikron	μ	1 μ = 1 μ m = 10^{-6} m

Clarification:

In table 8 are given the units out of SI and it is preferable that readers avoid them. If they are found in literature, they should first of all be transformed in SI units according to this table and after the achieved values are usable in SI.

Table 9 : Prefixes SI

Factor	Name	Symbol
10^{24}	Jota	Y
10^{21}	Zeta	Z
10^{18}	Ekza	E
10^{15}	Peta	P
10^{12}	Tera	T
10^9	Giga	G
10^6	mega	M
10^3	kilo	K
10^2	hekto	H
10^1	deka	da

Factor	Name	Symbol
10^{-1}	deci	d
10^{-2}	centi	c
10^{-3}	mili	m
10^{-6}	mikro	μ
10^{-9}	nano	n
10^{-12}	piko	p
10^{-15}	femto	f
10^{-18}	ato	a
10^{-21}	zepto	z
10^{-24}	jokto	y

Clarification:

SI prefixes are used to create decimal units. Decimal units are created from:

- all basic units of SI, except the unit of kilogram-kg (tab.2).
- all derived units SI especially named, except unit of degree celcius- $^{\circ}\text{C}$ (tab. 3).
- These units out of SI allowed to be used in special occasions and named especially: liter (L), tex (tex), bar (bar), elektronvolt (eV), and varn (var).

Table 10 : Prefixes for multiple binary

Name	Symbol	Factor
kibi	Ki	$2^{10} = 1024$
mebi	Mi	$2^{20} = 1\,048\,576$
gibi	Gi	$2^{30} = 1\,073\,741\,824$
tebi	Ti	$2^{40} = 1\,099\,511\,627\,776$
pebi	Pi	$2^{50} = 1\,125\,899\,906\,842\,624$
eksbi	Ei	$2^{60} = 1\,152\,921\,504\,606\,846\,975$
zebi	Zi	$2^{70} = 1\,180\,591\,620\,717\,411\,303\,424$
jobi	Yi	$2^{80} = 1\,208\,925\,819\,614\,629\,174\,706\,176$

Clarification:

In December 1998 **IEC (International Electrotechnical Commission)**, has approved standard names and symbols of prefixes for multiple binary. These prefixes are given in table 7. The names are formed from two first letters of the international prefix for multiples (*ki, me, gi, ti, pi, ex, ze, yo*), from letters bi that means *binary* and that can be applied in unit bit or byte. The symbols of binary multiples are formed from the symbol of international multiples– SI (K, M, G, T, P, E, Z, Y) and the letter *i* (Ki, Mi, Gi, Ti, Pi, Ei, Zi, Yi). In this case the multiple kilo got the symbol K and it can also be k.

Table 11: Comparisons with prefixes SI

a kibibit	1 Kibit = 2^{10} bit = 1 024 bit
a kilobit	1 kbit = 10^3 bit = 1 000 bit
a mebibajt	1 MiB = 2^{20} B = 1 048 576 B
a megabajt	1 MB = 10^6 B = 1 000 000 B
a gibibajt	1 GiB = 2^{30} B = 1 073 741 824 B
a gigabajt	1 GB = 10^9 B = 1 000 000 000 B

Clarification:

In table 11 are some examples of comparisons of binary units with international units.